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Trans. of Kogyo Kagaku Zasshi (Japan) 1953, v. 56, no. 7, p. 554-556.

We measured the viscosities of aniline hydrochloride-formalin water solutions and discovered the following facts. When the concentration of formaldehyde relative to aniline was increased progressively, emulsification took place and the solution solidified subsequently to a jelly-like state. At approximately the time that emulsification could be observed by the naked eye, the viscosity of the solution rose abruptly. When we held the quantity of formaldehyde relative to aniline constant and investigated the influence of the glass wall at 50, 40, 30, 20, and 10°C, it was seen to have no influence. The activation energy was calculated from the curve of (Materials--Plastics, TT, v. 5, no. 12) (over)

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1. Anilines--Chemical reactions
2. Formaldehyde--Chemical reactions
3. Resins--Viscosity

- I. Noda, M.
- II. Title: Studies...

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no. 3, p. 198-200.

We synthesized three systems of resins, whose proportions of formaldehyde relative to 1 M of aniline were 0.5 M, 0.7 M, and 1.0 M, and discovered the following facts. We measured the mean molecular weights and viscosity changes and found that although the curves ascended in the beginning, they began to decrease gradually past a certain point. Fractional precipitation was conducted with respect to the aniline resin of 1 : 0.7 mole ratio and its integral distribution curve and mass distribution curve were sought: it was seen as a result that the integral distribution curve stood up progressively in resins of after 45 (Materials--Plastics, TT, v. 5, no. 12) (over)

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1. Anilines--Chemical reactions
 2. Formaldehyde--Chemical reactions
 3. Resins--Molecular weight
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II. Title: Studies...

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The detection and determination of free amino groups
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2. Formaldehyde--Chemical
reactions
3. Resins--Molecular
structure
- I. Noda, M.
- II. Imoto, M.
- III. Title: Studies...

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- I. Crystallites
- I. Noda, T.
- II. Natsume, I.
- III. Title: Size...
- IV. ATS-64N49J
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